



## RESEARCH ARTICLE

## Synthesis, FT-IR analysis and DFT studies of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran

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### Abstract

In the present study 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran was prepared from 1-(N-nitroamidino)-3-ethoxyphenyl thiourea and 2-bromo acetylbenzofuran in the presence of triethylamine and characterized by FT-IR spectra. The geometry of the molecule was investigated and optimized with the help of B3LYP/6-31G DFT method using Gaussian 09 software package. The vibrational frequencies which were determined experimentally from spectral data are compared with those obtained theoretically *Ab-initio* and DFT calculations. The calculated HOMO-LUMO energy gap reveals that charge transfer occurs within the molecule.

### Keywords

Thiourea

FT-IR

DFT

Ab-initio

Homo-Lumo

### Introduction

Heterocyclic compounds have been an interesting field for a long time in medicinal chemistry. A Heterocyclic compounds containing nitrogen and sulphur atom serve as unique and versatile scaffolds for experimental drug design [1]. Benzothiazole consists of thiazole ring fused with benzene ring and posses, multiple applications, 2-aminobenzothiazoles were intensively studied as central muscle relaxants with glutamate neuro transmission in biochemical electro physiological and behavioral experiments benzothiazole ring [2] made

thiazole ring fused with benzene ring. Benzothiazole derivatives have been studied and found to have various chemical reactivity and biological activity such as antiviral, anti bacterial, antimicrobial activity [3]. Benzofuran derivatives posses a wide range of biological activities such as antimicrobial, antitumor and anti inflammatory activity [4].

### Experimental

The reagents and solvents used were of AR grade. All chemicals were purchased from Merck and Sigma Aldrich.

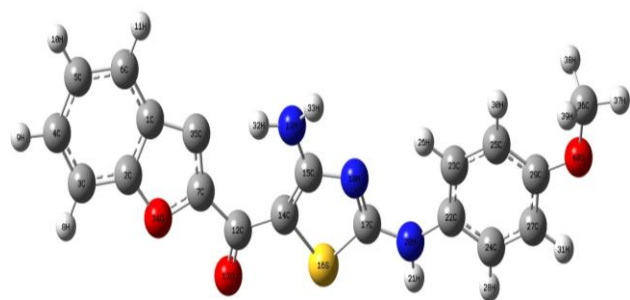
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### Synthesis of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran

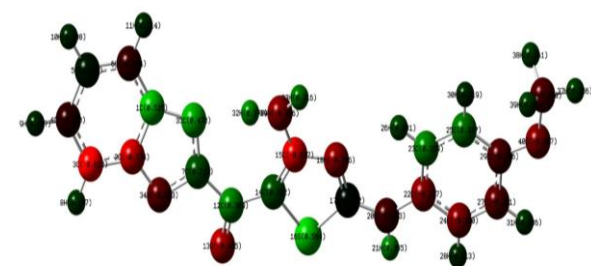
A solution of 1-aryl/alkyl-3-N-nitroamidino thiourea (1 mmol) in DMF (2 ml) was added to a solution of 2-bromoacetylbenzo furan (1 mmol) in DMF (2 ml). The reaction mixture was stirred well and triethylamine (1 mmol) was added. The reaction mixture was warmed at 80 – 85 °C for 15 min. It was then cooled and poured in to ice-cold water with constant stirring. A yellow orange precipitate thus obtained was filtered washed with water and dried. The crude product was recrystallized from methanol : water (2:1) ratio to give a yellow orange crystalline solid.

### Results and Discussion

Theoretical computations of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran have been performed using the B3LYP level of theory supplemented with standard 6-31G basis set using the Gaussian 09 software package [5]. Geometries have been first optimized structural parameters were used in the vibration frequency calculation at DFT level. The optimized structure is given in **Fig 1**.



**Fig 1** Optimized structure of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran.



**Fig 2** Mulliken atomic charges structure of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran.

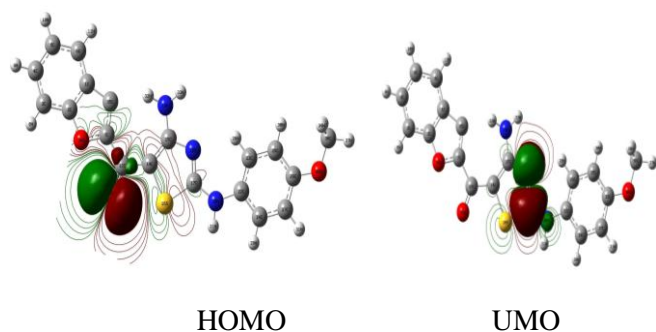
**Table 1** Mulliken atomic charges

Atom	Mulliken charges	Atom	Mulliken charges
C <sub>1</sub>	0.564	H <sub>21</sub>	0.353
C <sub>2</sub>	-0.574	C <sub>22</sub>	-0.610
C <sub>3</sub>	-0.816	C <sub>23</sub>	0.875
C <sub>4</sub>	-0.093	C <sub>24</sub>	-0.720
C <sub>5</sub>	0.056	C <sub>25</sub>	0.333
C <sub>6</sub>	-0.126	H <sub>26</sub>	0.209
C <sub>7</sub>	0.117	C <sub>27</sub>	-0.272
H <sub>8</sub>	0.148	H <sub>28</sub>	0.114
H <sub>9</sub>	0.109	C <sub>29</sub>	-0.309
H <sub>10</sub>	0.109	H <sub>30</sub>	0.131
H <sub>11</sub>	0.113	H <sub>31</sub>	0.142
C <sub>12</sub>	0.312	H <sub>32</sub>	0.391
O <sub>13</sub>	-0.466	H <sub>33</sub>	0.320
C <sub>14</sub>	0.380	O <sub>34</sub>	-0.266
C <sub>15</sub>	-0.627	C <sub>35</sub>	0.507
S <sub>16</sub>	0.540	C <sub>36</sub>	-0.250
C <sub>17</sub>	0.105	H <sub>37</sub>	0.171
N <sub>18</sub>	-0.388	H <sub>38</sub>	0.137
N <sub>19</sub>	-0.390	H <sub>39</sub>	0.182
N <sub>20</sub>	-0.138	O <sub>40</sub>	-0.376

### HOMO–LUMO

The optimized geometry of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran at DFT/B3LYP/6-31G level is shown in **Fig 3**. The HOMO–LUMO energies is calculated values of HOMO (-0.25048 a.u), and LUMO (-0.06435 a.u) and energy gap ( $\Delta E$ ) is (-0.18613 a.u),  $\Delta E$  reveals the

chemical activity of the molecule. LUMO as an electron acceptor represents the ability to obtain an electron and HOMO represents the ability to donate an electron.



**Fig 3** HOMO and LUMO of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran based on B3LYP/6-31G basis set

### Molecular geometry

2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran was subjected to optimization in the ground state. The optimized structural parameters were calculated by DFT/ B3LYP/6-31G basis set are listed in **Tables 1-3** in accordance with the atom numbering scheme given in **Fig 1**. The bond length of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran C<sub>1</sub>-C<sub>2</sub>, C<sub>2</sub>-C<sub>3</sub>, C<sub>3</sub>-C<sub>4</sub>, C<sub>4</sub>-C<sub>5</sub>, C<sub>5</sub>-C<sub>6</sub>, C<sub>7</sub>-C<sub>12</sub>, C<sub>12</sub>-C<sub>14</sub>, C<sub>14</sub>-C<sub>15</sub>, C<sub>22</sub>-C<sub>24</sub>, and C<sub>23</sub>-C<sub>25</sub> shows double character. The bond length of C<sub>7</sub>-C<sub>35</sub> is more than C<sub>7</sub>-C<sub>12</sub>; this is due to C<sub>7</sub>-C<sub>12</sub> double bond character, and C<sub>7</sub>-C<sub>35</sub> single bond character. The bond angle C<sub>14</sub>-S<sub>16</sub>-C<sub>17</sub> (88.287) is very less than C<sub>17</sub>-N<sub>20</sub>-C<sub>22</sub> (130.617) which is due to the fact that electro negativity of nitrogen is greater than sulphur. The dihedral angles show that the molecule consists of two planes.

**Table 2** Bond length of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran.

Atom	Bond length	Atom	Bond length
C <sub>1</sub> -C <sub>2</sub>	1.415	C <sub>2</sub> -C <sub>7</sub>	2.214
C <sub>2</sub> -C <sub>3</sub>	1.394	C <sub>3</sub> -H <sub>8</sub>	1.084
C <sub>3</sub> -C <sub>4</sub>	1.394	C <sub>4</sub> -H <sub>9</sub>	1.085
C <sub>4</sub> -C <sub>5</sub>	1.411	C <sub>5</sub> -H <sub>10</sub>	1.085
C <sub>5</sub> -C <sub>6</sub>	1.390	C <sub>6</sub> -H <sub>11</sub>	1.085

Atom	Bond length	Atom	Bond length
C <sub>7</sub> -C <sub>12</sub>	1.484	C <sub>24</sub> -C <sub>27</sub>	1.386
C <sub>12</sub> -O <sub>13</sub>	1.240	C <sub>24</sub> -H <sub>28</sub>	1.087
C <sub>12</sub> -C <sub>14</sub>	1.443	C <sub>25</sub> -C <sub>29</sub>	1.397
C <sub>14</sub> -C <sub>15</sub>	1.402	C <sub>25</sub> -H <sub>30</sub>	1.083
C <sub>14</sub> -S <sub>16</sub>	1.776	C <sub>27</sub> -H <sub>31</sub>	1.084
S <sub>16</sub> -C <sub>17</sub>	1.759	N <sub>19</sub> -H <sub>32</sub>	1.012
C <sub>17</sub> -N <sub>18</sub>	1.316	N <sub>19</sub> -H <sub>33</sub>	1.009
C <sub>15</sub> -N <sub>19</sub>	1.363	C <sub>2</sub> -O <sub>34</sub>	1.366
C <sub>17</sub> -N <sub>20</sub>	1.358	C <sub>7</sub> -C <sub>35</sub>	1.359
N <sub>20</sub> -H <sub>21</sub>	1.010	C <sub>29</sub> -C <sub>36</sub>	2.396
N <sub>20</sub> -C <sub>22</sub>	1.414	C <sub>36</sub> -H <sub>37</sub>	1.090
C <sub>22</sub> -C <sub>23</sub>	1.398	C <sub>36</sub> -H <sub>38</sub>	1.097
C <sub>22</sub> -C <sub>24</sub>	1.407	C <sub>36</sub> -H <sub>39</sub>	1.097
C <sub>23</sub> -C <sub>25</sub>	1.400	C <sub>29</sub> -O <sub>40</sub>	1.368
C <sub>23</sub> -H <sub>26</sub>	1.080		

**Table 3** Bond angle of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran.

Atom	Bond angle	Atom	Bond angle
C <sub>1</sub> -C <sub>2</sub> -C <sub>3</sub>	122.90	C <sub>2</sub> -C <sub>7</sub> -C <sub>12</sub>	151.21
C <sub>2</sub> -C <sub>3</sub> -C <sub>4</sub>	116.63	C <sub>7</sub> -C <sub>12</sub> -O <sub>13</sub>	118.23
C <sub>3</sub> -C <sub>4</sub> -C <sub>5</sub>	121.53	C <sub>7</sub> -C <sub>12</sub> -C <sub>14</sub>	121.51
C <sub>4</sub> -C <sub>5</sub> -C <sub>6</sub>	121.35	C <sub>12</sub> -C <sub>14</sub> -C <sub>15</sub>	137.47
C <sub>1</sub> -C <sub>2</sub> -C <sub>7</sub>	74.753	C <sub>12</sub> -C <sub>14</sub> -S <sub>16</sub>	113.82
C <sub>2</sub> -C <sub>3</sub> -H <sub>8</sub>	121.023	C <sub>14</sub> -S <sub>16</sub> -C <sub>17</sub>	88.28
C <sub>3</sub> -C <sub>4</sub> -H <sub>9</sub>	119.21	S <sub>16</sub> -C <sub>17</sub> -N <sub>18</sub>	115.92
C <sub>4</sub> -C <sub>5</sub> -H <sub>10</sub>	119.14	C <sub>14</sub> -C <sub>15</sub> -N <sub>19</sub>	128.02
C <sub>5</sub> -C <sub>6</sub> -H <sub>11</sub>	121.17	S <sub>16</sub> -C <sub>17</sub> -N <sub>20</sub>	118.94

Bond angle	Value (°)	Bond angle	Value (°)
C <sub>17</sub> -N <sub>20</sub> -H <sub>21</sub>	115.04	C <sub>24</sub> -C <sub>27</sub> -H <sub>31</sub>	120.86
C <sub>17</sub> -N <sub>20</sub> -C <sub>22</sub>	130.61	C <sub>15</sub> -N <sub>19</sub> -H <sub>32</sub>	119.68
N <sub>20</sub> -C <sub>22</sub> -C <sub>23</sub>	124.29	C <sub>15</sub> -N <sub>19</sub> -H <sub>33</sub>	115.84
N <sub>20</sub> -C <sub>22</sub> -C <sub>24</sub>	116.79	C <sub>1</sub> -C <sub>2</sub> -O <sub>34</sub>	111.55
C <sub>22</sub> -C <sub>23</sub> -C <sub>25</sub>	120.15	C <sub>2</sub> -C <sub>7</sub> -C <sub>35</sub>	72.00
C <sub>22</sub> -C <sub>23</sub> -H <sub>26</sub>	119.71	C <sub>25</sub> -C <sub>29</sub> -C <sub>36</sub>	93.52
C <sub>22</sub> -C <sub>24</sub> -C <sub>27</sub>	120.91	C <sub>29</sub> -C <sub>36</sub> -H <sub>37</sub>	135.97
C <sub>22</sub> -C <sub>24</sub> -H <sub>28</sub>	119.89	C <sub>29</sub> -C <sub>36</sub> -H <sub>39</sub>	95.22
C <sub>23</sub> -C <sub>25</sub> -C <sub>29</sub>	120.65	C <sub>25</sub> -C <sub>29</sub> -O <sub>40</sub>	124.98
C <sub>23</sub> -C <sub>25</sub> -H <sub>30</sub>	118.33		

**Table 4** Dihedral angle of bond length of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-yl)benzofuran.

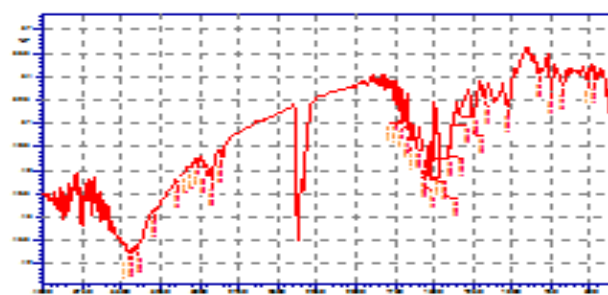
Atom	Dihedral angle	Atom	Dihedral angle
C <sub>1</sub> -C <sub>2</sub> -C <sub>3</sub> -C <sub>4</sub>	0.612	C <sub>12</sub> -C <sub>14</sub> -C <sub>15</sub> -N <sub>19</sub>	-5.657
C <sub>2</sub> -C <sub>3</sub> -C <sub>4</sub> -C <sub>5</sub>	-0.248	C <sub>14</sub> -S <sub>16</sub> -C <sub>17</sub> -N <sub>20</sub>	-179.68
C <sub>3</sub> -C <sub>4</sub> -C <sub>5</sub> -C <sub>6</sub>	-0.178	S <sub>16</sub> -C <sub>17</sub> -N <sub>20</sub> -H <sub>21</sub>	-0.618
C <sub>6</sub> -C <sub>1</sub> -C <sub>2</sub> -C <sub>7</sub>	179.67	S <sub>16</sub> -C <sub>17</sub> -N <sub>20</sub> -C <sub>22</sub>	179.54
C <sub>1</sub> -C <sub>2</sub> -C <sub>3</sub> -H <sub>8</sub>	-179.54	C <sub>17</sub> -N <sub>20</sub> -C <sub>22</sub> -C <sub>23</sub>	-1.869
C <sub>2</sub> -C <sub>3</sub> -C <sub>4</sub> -H <sub>9</sub>	179.71	C <sub>17</sub> -N <sub>20</sub> -C <sub>22</sub> -C <sub>24</sub>	178.36
C <sub>3</sub> -C <sub>4</sub> -C <sub>5</sub> -H <sub>10</sub>	179.86	N <sub>20</sub> -C <sub>22</sub> -C <sub>23</sub> -C <sub>25</sub>	179.76
C <sub>4</sub> -C <sub>5</sub> -C <sub>6</sub> -H <sub>11</sub>	179.98	N <sub>20</sub> -C <sub>22</sub> -C <sub>23</sub> -H <sub>26</sub>	0.214
C <sub>1</sub> -C <sub>2</sub> -C <sub>7</sub> -C <sub>12</sub>	-173.80	N <sub>20</sub> -C <sub>22</sub> -C <sub>24</sub> -C <sub>27</sub>	179.76
C <sub>2</sub> -C <sub>7</sub> -C <sub>12</sub> -O <sub>13</sub>	7.529	N <sub>20</sub> -C <sub>22</sub> -C <sub>24</sub> -H <sub>28</sub>	-0.2728
C <sub>2</sub> -C <sub>7</sub> -C <sub>12</sub> -C <sub>14</sub>	-172.84	C <sub>22</sub> -C <sub>23</sub> -C <sub>25</sub> -C <sub>29</sub>	0.0154
C <sub>7</sub> -C <sub>12</sub> -C <sub>14</sub> -C <sub>15</sub>	6.923	C <sub>22</sub> -C <sub>23</sub> -C <sub>25</sub> -H <sub>30</sub>	179.95
C <sub>7</sub> -C <sub>12</sub> -C <sub>14</sub> -S <sub>16</sub>	-177.55	C <sub>22</sub> -C <sub>24</sub> -C <sub>27</sub> -H <sub>31</sub>	-179.99
C <sub>12</sub> -C <sub>14</sub> -S <sub>16</sub> -C <sub>17</sub>	-177.14	C <sub>14</sub> -C <sub>15</sub> -N <sub>19</sub> -H <sub>32</sub>	15.79
C <sub>14</sub> -S <sub>16</sub> -C <sub>17</sub> -N <sub>18</sub>	0.325	C <sub>14</sub> -C <sub>15</sub> -N <sub>19</sub> -H <sub>33</sub>	165.84

Atom	Dihedral angle	Atom	Dihedral angle
C <sub>6</sub> -C <sub>1</sub> -C <sub>2</sub> -O <sub>34</sub>	179.68	C <sub>25</sub> -C <sub>29</sub> -C <sub>36</sub> -H <sub>38</sub>	55.05
C <sub>1</sub> -C <sub>2</sub> -C <sub>7</sub> -C <sub>35</sub>	1.15	C <sub>25</sub> -C <sub>29</sub> -C <sub>36</sub> -H <sub>39</sub>	55.04
C <sub>23</sub> -C <sub>25</sub> -C <sub>29</sub> -C <sub>36</sub>	179.96	C <sub>23</sub> -C <sub>25</sub> -C <sub>29</sub> -O <sub>40</sub>	179.99
C <sub>25</sub> -C <sub>29</sub> -C <sub>36</sub> -H <sub>37</sub>	179.97		

### Infrared Spectra

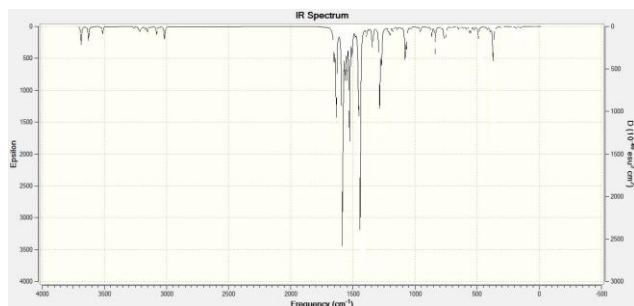
The compound 2-(4-amino-2-(4-ethoxy phenylamino)thiazol-5-yl)benzofuran was obtained in 98% yield with melting point 155-158 °C. Based on the elemental analysis the molecular composition of the compound was found to be C<sub>19</sub>H<sub>14</sub>N<sub>3</sub>O<sub>3</sub>S. The IR (KBr) spectrum of the compound shows N-H peaks at 3454, 3425, and 3400 cm<sup>-1</sup> respectively. The aromatic C-H bond gives peak at 3078 cm<sup>-1</sup>, 3045 cm<sup>-1</sup> 3078 cm<sup>-1</sup>. The carbonyl group which occurred at 1649 cm<sup>-1</sup>.

The vibration study of the molecule is compared with the calculated data and is shown in **Fig 4**. The calculated band at 3685 cm<sup>-1</sup> is due to asymmetric N-H stretching of N<sub>19</sub>-H<sub>32</sub> and N<sub>19</sub>-H<sub>33</sub> but experimentally it is assigned at 3454 cm<sup>-1</sup>. Calculated band at 3220 cm<sup>-1</sup> is due to asymmetric C-H stretching of C<sub>23</sub>-H<sub>26</sub> and N<sub>20</sub>-H<sub>21</sub> calculated band at 3078 cm<sup>-1</sup>, is due to asymmetric C-H stretching of C<sub>23</sub>-H<sub>26</sub>, C<sub>25</sub>-H<sub>30</sub>, C<sub>27</sub>-H<sub>31</sub>, C<sub>24</sub>-H<sub>28</sub>. The C=C stretching is experimentally observed around 1562 cm<sup>-1</sup> which is close agreement with the calculated frequency 1566, 1550, 1514 and 1502 cm<sup>-1</sup>. The bands with in the range 1585-1015 cm<sup>-1</sup> are assigned for C-H in plane bending vibrations which is in close agreement with the experimental values.



a) Experimental

**Fig 4** Experimental and theoretical FT-IR spectrum of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-yl)benzofuran.



### b) Theoretical

## Conclusions

2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran has been synthesized and characterized by FT-IR spectrum. The structure of 2-(4-amino-2-ethoxyphenylamino thiazol-5oyl)benzofuran was optimized by DFT method using the basis set B3LYP/6-31G. The HOMO and LUMO energy gap reveals the activity of 2-(4-amino-2-(4-ethoxyphenylamino)thiazol-5-oyl)benzofuran.

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